

Europäisches Patentamt
European Patent Office
Office européen des brevets



(11) **EP 1 146 057 A1**

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication:
17.10.2001 Bulletin 2001/42

(21) Application number: 00108313.8

(22) Date of filing: 15.04.2000

(51) Int Cl.7: **C08F 2/44**, C11D 17/00,
A61K 7/46, A61K 7/48,
A61K 7/06, A61K 7/00,
C11D 3/50

(84) Designated Contracting States:
AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE
Designated Extension States:
AL LT LV MK RO SI

(71) Applicant: **Givaudan SA**
1214 Vernier-Genève (CH)

(72) Inventors:
• **Quellet, Christian**
2502 Biel (CH)

• **Hotz, Jutta**
4058 Basel (CH)
• **Balmer, Marc**
8103 Unterengstringen (CH)

(74) Representative: **Patentanwälte**
Schaad, Balass, Menzi & Partner AG
Dufourstrasse 101
Postfach
8034 Zürich (CH)

(54) **Polymeric nanoparticles including olfactive molecules**

(57) The polymeric nanoparticles including olfactive molecules have a glass temperature of $>50^{\circ}\text{C}$. They are obtainable by continuously adding a monomer component and an olfactive component to an aqueous solution. The aqueous solution comprises a first initiator and an emulsifier. The added components are distributed in the aqueous solution to obtain reaction mixture while the polymerization of the monomer component is started at

the first temperature. The liquid monomer component and the olfactive component are added continuously while a second initiator is added dropwise to the reaction mixture and the first temperature is maintained. After terminating the addition of liquid monomer component and the olfactive component the temperature of the reaction mixture is increased to a second temperature and a third initiator is added dropwise.

EP 1 146 057 A1

Description

[0001] The present invention relates to polymeric nanoparticles including olfactive molecules which are useful as delivery system for olfactive compounds.

[0002] Olfactive compound in the present context means fragrances, odour masking agents and mixtures thereof such as perfume compositions as well as precursors for the above.

[0003] A principal strategy currently employed in imparting odours to consumer products is the admixing of the fragrance directly into the product. There are, however, several drawbacks to this strategy. The fragrance material can be too volatile, resulting in fragrance loss during manufacturing, storage and use. Many fragrance materials are also unstable over time. This again results in loss of perfume during storage.

[0004] In many consumer products it is desirable for the fragrance to be released slowly over time. Since the most volatile fragrances, or "top notes" are responsible for the "fresh feeling" consumers experience, it is desirable that in addition to the less volatile fragrances of a fragrance composition also the more volatile fragrances are slowly released. Such slow release is commonly referred to as sustained release and is characterized, ideally, by a zero-order diffusion kinetics. For olfactive compositions, the signature of a zero-order diffusion kinetics is a constant value of the fragrance concentration in the gas phase in the vicinity of the substrate over a long period of time. In the language of perfumery, the gas phase around any substrate is called "headspace" and, accordingly, the fragrance concentration in this phase is called "headspace concentration". The headspace concentration is usually expressed in ng/l headspace.

[0005] To be perceived by the consumer, the fragrance headspace concentration must be higher than the so-called olfactive threshold, being the concentration below which the human nose perceives. Hence, long lasting means that the headspace concentration is maintained above the threshold concentration over a long period of time.

[0006] The olfactory performance of fragrances can alternatively be expressed in terms of "odour value". The odour value is the ratio of the actual headspace concentration (vapour pressure) of the fragrance to its olfactory threshold value, both expressed in ng/l. Odour values can be represented on a vapour pressure vs. olfactory threshold log-log map. Fragrances with identical odour values are found on straight lines on the log-log map.

[0007] Fragrances that are considered as top notes are located on the upper side realm of the log-log map. Fragrances considered as top notes and those having odour values between 10'000 and 10'000'000 are preferred in the present invention.

[0008] Direct admixing of a fragrant composition in a product does generally not provide the desired zero-order release kinetics quoted above. Instead, a first order release kinetics is observed, which is marked by an exponential decrease of the headspace concentration. The kinetics of this first-order diffusion depends on the vapour pressure of the fragrance. Due to this vapour pressure dependence, top notes are perceived only over a very short period.

[0009] Therefore, it is highly desirable to provide a system that (i) protects the volatile top notes in a composition, (ii) improves the substantivity of these top notes on various substrates and (iii) imparts a long-lasting release of these top notes from various substrates over a long period of time, i.e. at least 24 hours, preferably at least 72 hours and more preferably at least 5 days.

[0010] A large amount of work has been devoted to the realization of fragrance delivery systems that expand the spectrum of fragrance perceived on substrates and improve the sustained release of fragrances.

[0011] Micro-encapsulation and inclusion complexes with cyclodextrines have been used to decrease volatility, improve stability and provide slow-release properties. However, cyclodextrines poured into water release the fragrance immediately, which limits their uses as controlled release systems in e.g. laundry care.

[0012] Spray drying, coacervation and interfacial polymerization are examples of technologies which are also used for encapsulating fragrances in detergents and laundry care products. Whereas these techniques may increase the amount of fragrances available in the washing liquor, they do not improve the sustained release of fragrance from substrates per se.

[0013] An important part of the fragrance composition initially added to detergent products, soaps and conditioners is lost by evaporation and dissolution in the wash and rinse water. Accordingly, an important amount of work has been devoted to the general problem of perfuming various substrates during washing and thereafter during the conditioning procedure in order to enhance fragrance deposition on the substrates and to provide a sustained release of fragrance during and after the drying stage.

[0014] A classical method for enhancing fragrance deposition is admixing the fragrance with surfactants and especially with cationic surfactants contained in conditioner compositions. An alternative route involves the preparation of solid particles by admixing the fragrance with amphiphilic polymers. EP-A 0 469 228 discloses a perfume carrier and delivery system which is claimed to be suitable for use in a wash liquor and which consists of a perfume, a hydrophobic solid, a branched or linear alcohol or ester and an amphiphilic polymer derived from monomers such as ethylene oxide, acrylic acid, styrenesulfonate, acrylamide and block copolymers thereof with polyoxypropylene or polyamide.

[0015] EP-A 0 346 034 discloses a process for preparing particles of wax encapsulated actives (like perfume) for use in cleaning products by (i) dispersing actives in molten wax, (ii) emulsifying the active/wax dispersion in aqueous

surfactant solution, (iii) quenching the capsules by cooling and (iv) retrieving solidified capsules.

[0016] US-A 5,506,201 discloses a method for producing a fragrance-containing solid particle of improved substantivity for incorporation into laundry detergents which comprises of a fat component and a solid surface active agent like sorbitan ester.

[0017] The methods of the above documents may improve the deposition of fragrance onto substrates when added to laundry care products. However, it is known that waxes and fats exude hydrophobic volatile molecules and therefore they do not reduce the evaporation rate of the fragrances from the substrate during drying. Thus, these systems do generally not produce the desired sustained and controlled release action.

[0018] EP-A 0 617 051 discloses polymeric composition obtained by emulsion polymerization of unsaturated monomers in the presence of a fragrance. The composition is used for fragrant coatings for artefacts, however, in neither application of these latexes. No details concerning the method to produce this polymer composition is disclosed.

[0019] JP-A 63/122796 discloses the use of a non fragrant latex in liquid products, such as laundry care products, comprising fragrances, to improve deposition and controlled release of fragrances on substrates. The polymer used is soft polyacrylate having low glass temperature bearing cationic or pseudo-cationic groups, and consisting essentially of polybutylacrylate.

[0020] WO 98/28396 discloses a latex having particle sizes larger than 1 to 10 microns which are produced by suspension polymerization of vinyl monomers. A fragrance is incorporated by mixing it with the monomers prior to polymerization or post-addition of the fragrance to the latex. The latex microparticles are surrounded by a polyhydroxy hydrocolloid layer which is claimed to enhance deposition of the perfume on substrates.

[0021] All the known systems may potentially improve perfume deposition on substrates but do not impart sustained, controlled release of volatile top notes to the substrates.

[0022] EP-A 0 925 776 discloses a polymer molecularly imprinted, with an organoleptic substance and prepared by solution or dispersion, polymerisation in organic media, in the presence of the imprinting substance. The polymer used binds preferentially the olfactive compounds which have been used for imprinting. Examples of selective malodour recognition are given and some of the results with laundry care products suggest a better deposition and enhanced longevity of imprinting agents. There are, however, a number of drawbacks. First, solution polymerization is difficult to conduct at industrial scales. Second, subsequent grinding and sieving to obtain usable particles is required. Third, the use of an organic solvent is detrimental to the final loading of the polymer, due to partition of the olfactive compound between polymer and solvent.

[0023] One object of the present invention is, therefore, to provide a novel polymeric system for delivery of olfactive compounds or components and precursors thereof. In the following the term olfactive compounds or components also encompasses masking agents and precursors for olfactive compounds and masking agents.

[0024] A further object of the present invention is to provide polymeric nanoparticles for the delivery and uptake of olfactive components.

[0025] A further object of the present invention is to provide polymeric nanoparticles with enhanced substantivity to substrates and controlled sustained release for the olfactive components.

[0026] A special object of the present invention is to provide a new delivery system for top notes.

[0027] A further object of the present invention is to provide a new delivery system for olfactive components with sustained release at room temperature and with fast or flash release at high ironing or dryer temperature.

[0028] A further object of the invention is to provide a novel delivery system for olfactive compositions comprising different fragrant molecules with significantly improved sustained release of the entire olfactive composition and particularly of the most volatile top notes during and after application.

[0029] A further object of the present invention is to increase the olfactive performance of detergents, fabric softeners, body care and home care products and cosmetics with respect to perfuming of fabrics, solid surfaces, hair and skin.

[0030] The present invention relates to polymeric nanoparticles including olfactive molecules, having the above characteristics and a glass temperature of $>50^{\circ}\text{C}$ are obtainable by

a) continuously adding a liquid monomer component and an olfactive component to an aqueous solution of a first initiator comprising an emulsifier and distributing the added components in the aqueous solution to obtain a reaction mixture while starting polymerisation of the monomer component in the reaction mixture at a first temperature and, while continuing the addition of the liquid monomer component and the olfactive component, adding a second initiator dropwise to the reaction mixture while maintaining the first temperature

b) after terminating the addition of the liquid monomer component and the olfactive component increasing the temperature of the reaction mixture to a second temperature and dropwise adding a third initiator.

[0031] It has to be emphasised that the addition of the monomer component and the olfactive component has to be continued during the whole step a).

[0032] It is assumed that under these conditions the polymer particles of the invention are imprinted by the molecules of the olfactive composition, i.e. the polymer is formed around the individual olfactive molecules. Such polymers are also called template polymers. (Molecular) imprinting is characterised by (i) an enhanced retention of the olfactive molecules in the particles and (ii) a reduced diffusion of these olfactive molecules through the polymer. The overall retention/release profile of (molecularly) imprinted polymers is suitable to satisfy the objects of the present invention. Polymers obtained by batch radical polymerization, where the whole monomer component and olfactive component is present in the reaction mixture before polymerization starts, do not satisfy the objects of the present invention. Similarly, the process described above yields nanoparticles that perform better in their controlled release behaviour than perfumed nanoparticles obtained by absorbing an olfactive component into pre-formed nanoparticles. It has been surprisingly found that the olfactive molecules do generally not suffer from the polymerization conditions. It is also surprising that template polymers can be obtained with the generally small and highly volatile molecules of top notes.

[0033] Further it is also surprising that the formation of imprinted or template polymers according to the invention do not require the addition of large amounts of crosslinking agents in the reaction mixture.

[0034] An additional advantage of the polymeric nanoparticle of the invention is the extremely low amount of monomers found. The amount of the principal monomer component is found in an amount of 100 ppm or less. The principal monomer component being added during the process of the invention in a significantly higher amount than the other monomer(s).

[0035] The above process is also called semicontinuous batch polymerization.

[0036] The polymerization takes place in a two phase system and with homogenous distribution of the liquid monomer and the olfactive composition in the aqueous solution of the emulsifier is important. The two phase polymerization is either a dispersion, suspension or preferably an emulsion or mini-emulsion polymerization. For the latter a pre-emulsion is prepared by admixing the monomer with the olfactive composition and adding the pre-emulsion during step a) to the aqueous solution.

[0037] Optionally the aqueous solution may comprise dispersed polymer seeds.

[0038] Depending on the desired particle size, size distribution, fragrance release performance, etc., the steps of the above process can be modified. The rate of the addition and the droplet size of the olfactive and the monomer component and the initiator can be varied to obtain the desired particle properties. In particular, step a) can be followed by the addition of additional monomers and initiators, in order to provide an outer layer or coating of desired character on the particles. Also, the composition of the monomer and the olfactive component can be continuously varied in order to provide a gradient of characteristics within the particles. Such variations include: (i) variation of the temperatures, (ii) changing one or more initiators and/or initiator concentrations, (iii) changing feed rate of the monomer and olfactive component, (iv) changing the period of the above steps.

[0039] The composition of the monomer is particularly relevant to the particle performance. Monomer yielding polymers with a high glass transition temperature are particularly suitable. These encompass for example, styrene, methyl (meth)acrylate, isobornyl(meth)acrylate, adamantyl (meth)acrylate and (meth)acrylic acid, acrylamide and monomers of the general formula $=C(R^1)-CO-(CH_2)_n-X^+ Y^-$, wherein R^1 is H or CH_3 ; n is 1 or 2, X is either a trimethyl quaternary ammonium or a dimethyl sulfonium radical and Y^- is a counterion.

[0040] Preferred monomers are styrene, methyl(meth)acrylate and (meth)acrylic acids.

[0041] Cross linking monomers particularly useful for the present invention comprise divinyl benzene, trivinyl benzene, divinyl toluene, trivinyl toluene, di- and tri-acrylates like diesters - formed by (meth)acrylic acid and diols - and higher esters - formed by (meth)acrylic acid and polyols. Preferred are divinyl benzene, triethyleneglycol dimethacrylate, tetraethyleneglycol dimethacrylate, allylmethacrylate, diallylmalate, triallylmalate and 1,4-butanediol diacrylate.

[0042] The choice of emulsifier is less critical. Useful emulsifiers for emulsion and mini-emulsion polymerization can be anionic, cationic, zwitterionic or non-ionic. Examples of useful emulsifiers are given in Table 1:

Table 1

Trade Name	Common Name	Remarks
Rewoquat® RTM50	Ricinoylamidopropyltrimethyl -ammoniummetho sulfate	cationic
Rewoquat® CPEM	Cocopentylethoxymethyl-ammoniummetho sulfate	cationic
Ethoquat® C12	Cocobis(2-hydroxyethyl) methylammonium chloride	cationic
CTAB	Cetyltrimethylammonium bromide	cationic
Lexemul® AR	Glyceryl stearate (and) Stearamidoethyl diethylamine	cationic

EP 1 146 057 A1

Table 1 (continued)

Trade Name	Common Name	Remarks
Disponal® A1080	Mixture of ethoxylated linear fatty alcohols	nonionic
Disponal® A3065	Mixture of ethoxylated linear fatty alcohols	nonionic
Mergital® LM4L	Mixture of C ₁₂ -C ₁₃ fatty alcohols ethoxylated with 4 moles of ethylene oxide	nonionic
Lauropal® 12	Mixture of C ₁₂ -C ₁₄ fatty alcohols ethoxylated with 6-15 moles of ethylene oxide	nonionic
Montane® 60	Sorbitan stearate	nonionic
Tween® 20	Polysorbate 20	nonionic
Tween® 80	Polysorbate 80	nonionic
Tritons® X 165	Stearate	nonionic
Dowfax® 2A1	Disulfonated surfactant with tetrapropylene hydrophobe source	anionic
SDS	Sodium dodecyl sulfate	anionic
Abex® 3594		anionic
Abex® EP-227	Ammoniumnonoxynol-77 sulfate	anionic
Lexemul® AS	Glyceryl stearate (and) Sodium lauryl sulfate	anionic

[0043] Preferred are SDS, Abex® 3594 and Dowfax® 2A1 which are common emulsifiers for polystyrene particles.

[0044] Alternatively, polymer emulsifiers can be used, either as hydrocolloid stabilizing agents or as emulsifiers. Hydrocolloids of interest are poly(vinylalcohol-co-vinylacetate) copolymers, modified cellulose, polyoxyethylene and polyvinylpyrrolidon. Polymer surfactants are for example multiblock copolymers and graft copolymers containing at least one hydrophilic block and at least one hydrophobic block, like polyoxyethylene-polyoxypropylene-polyoxyethylene (PLURONIC) copolymers, polyether-modified dimethicones and polyether-alkyl-dimethicones (ABIL) copolymers. Cationic silicones and polymers containing polyimide moieties may be also useful.

[0045] Initiators useful for emulsion polymerization are watersoluble initiators like peroxodisulfates, organic peroxides, hydroperoxides and water soluble azo-compounds. Specific examples of suitable initiators are ammonium persulfate, sodium persulfate, potassium persulfate, 1,4-diisopropylbenzene hydroperoxide, cumene hydroperoxide, 2,2'-azobis(2-methylpropio-namidine)dihydrochlorid and 4,4'-azobis(4-cyanovaleic acid).

[0046] Preferred are the redox systems of ammonium- or sodium persulfates with iron (II) sulfate which allow thermic initiations at low temperatures.

[0047] Among the huge variety of fragrant materials used in perfumery, the following volatile compounds are particularly suitable as imprinting agents.

[0048] List of the olfactive components (OV: Odor value)

Table 2

Chemical Name	Odor Value (OV)
ETHYL 2-METHYL PENTANOATE	62586530
ETHYL ISO BUTYRATE FCC	39200330
BUT-2-ENE-1-ONE, 1-(2,6,6-TRIMETHYL-	21672290

	1,3-CYCLOHEXADIEN-1-YL)-	
5	NONADIENAL 1% CITROFLEX	17446920
	PHENYL ACETALDEHYDE 85	14342060
10	INDOLE, 3-METHYL	10531770
	HEXANOIC ACID, ETHYL ESTER	10296830
15	1,8 - EPOXY - PARA - METHANE	9931397
	ALDEHYDE C12 MNA	9648383
	ETHYL ACETATE	7672145
20	ETHYL-2-METHYL BUTYRATE	6716841
	ALDEHYDE C8	5678510
25	TRANS DECEN-4-AL	4668142
	UNDECATRIENE	4482573
30	METHYL OCTYL ACETALDEHYDE	4469274
	ALDEHYDE C7	3699238
	2,6 DIMETHYL-5-HEPTENAL	3674714
35	DIHYDRO 2-METHOXY-4-(2-PROPENYL) PHENOL	3472000
	BUTYL PHTHALIDE	3266946
40	LIME OXIDE	2161723
	ETHYL BUTYRATE	1764598
45	NONENOL, CIS-6	1712805
	PARA-TOLYL ALDEHYDE EXTRA	1475808
50	DIHYDRO ANETHOLE FCC	1313749
	CYCLO HEXYL ACETATE	1292485
55	PARA-CRESYL ACETATE	1273204

3	PYRAN, 2, 4-DIMETHYL-6-PHENYL-DIHYDRO-	1193400
	3-(3-ISOPROPYLPHENYL) BUTANAL	1187323
	BUT-2-ENE-1-ONE, 1-(2, 6, 6, - TRIMETHYLCYCLOHEX-1-EN-1-YL) -	1122995
10	METHYL 2-OCTYNOATE	1094551
	BUTYL ACETATE, ISO, FCC	979809.6
15	7-METHYL-3, 4-DIHYDRO-2H-1, 5- BENZODIOXEPIN-3-ONE	973365.4
	2-BUTEN-1-ONE, 1-(2, 6, 6-TRIMETHYL-1, 3- CYCLOHEXEN-1-YL) -	961551.6
20	2-METHOXY-4-METHYLPHENOL	943751.4
	4-METHOXYBENZALDEHYDE	941980.3
25	OCTENOL	925978.5
	BICYCLO[3.1.1]HEPT-2-ENE-2- PROPANAL, 6, 6-DIMETHYL	904490.6
30	ACETIC ACID, 3-METHYL-2-PROPENYL ESTER	877798.3
	ALDEHYDE ISO C11	854722
35	3, 7-DIMETHYL-2, 6-OCTADIENENITRILE	832569
	METHYL ANTHRANILATE	827264.3
40	METHYL OCTALACTONE	826643
	METHYL PHENYL ACETATE	792198.8
	METHYL IONONE / IRISANTHEME	778605.1
45	DIMETHYL ACETOPHENONE, 2, 4	754754.7
	8-ETHYL-1-OXASPIRO[4.5]DECAN-2-ONE	734286
50	4-METHOXYBENZALDEHYDE	683617.5
55	CYCLOHEX-2-EN-1-ONE, 2-METHYL-5-(1- METHYLETHENYL)	673272.8

EP 1 146 057 A1

5	PHENYL PROPYL ALDEHYDE	589314.7
	ALLYL AMYL GLYCOLATE	587664.6
10	2,4 DIMETHYL 3 CYCLOHEXEN-1-CARBALDEHYDE	587309.8
	CYCLOHEX-3-ENE, 1-CARBOXALDEHYDE, 2.4-DIMETHYL-	585468
15	HEXENYL ACETATE, CIS-3-	575465.8
	7-METHYL-3-METHYLENE-1,6-OCTADIENE	546499.8
	PARA CRESOL	524810.3
20	ISO NONYL ALDEHYDE	510654.2
	ALDEHYDE C18	507873.9
25	ALLYL AMYL GLYCOLATE	503042.8
	CYCLOPENT-2-EN-1-ONE, 2-HEXYL-	485812.6
	ISO PROPYL QUINOLINE / BASE 3	479565.3
30	2-PHENLYPROPANAL	477838.4
	AMYL ACETATE	441261.3
35	BUTYL ACETATE	435143.8
	METHYL LAITONE	420065.1
	ALDEHYDE C14	419891.6
40	HEXENOL, CIS-3-	391510
	ALDEHYDE C10	387743.2
45	BUTYRIC ACID, 2-METHYL-,HEXYL ESTER	382190.6
	ETHYL PHENYL ACETATE	375894.8
50	ACETIC ACID, (CYCLOHEXYLOXY)-, 2-PROPENYL ESTER	372893.1
	DEC-4-EN-1-AL	367957

55

5	METHYL HEXYL KETONE	361139.4
	FORMYL-7-ISOPROPYL-5-METHYL-2-BICYCLO[2.2.2.]OCT-2-ENE	355730.8
10	2,6 DIMETHYLHEPTAN-2-OL	352155.6
	ALDEHYDE C11 UNDECYLENIC	351095
	TRANS 2 HEXENAL	350064.3
15	2-METHOXY-4-(2-PROPENYL) PHENOL	346904.9
	3.7-DIMETHYLOCTANAL	311427.9
20	ETHYL HEPTOATE	310684.7
	3,7-DIMETHYL-2,6-OCTADIENENITRILE	298260.5
25	CYCLOHEX-1-ENE 4-CARBOXALDEHYDE-1,3,5-TRIMETHYL	295165.8
	PYRAN, TETRAHYDRO-4-METHYL-2-(2-METHYL-1-PROPENYL)-2H	291002
30	METHYL PHENYL ETHYL OXYDE	287117.9
	HEXENYL FORMATE, CIS-3-	282721.5
35	BUT-3-EN-2-ONE, 4-82,6,6-TRIMETHYLCYCLOHEX-1-EN-1-YL)	280385
	HEX-5-EN-2-ONE	278762.7
40	CARYOPHYLLENE FCC ,NATURAL NKFP	278378
	PARA METHYL ACETOPHENONE	271792.6
	JASVERATE	265515.9
45	NONA-6,8-DIEN-3-ONE, 2,4,4,7-TETRAMETHYL	252222
50	DIHYDRO 3,7 DIMETHYL-1,6-OCTADIEN-3-OL	250072.3
	OCT-7-EN-2-OL, 2,6-DIMETHYL	248873.6
55	2-BUTEN-1-ONE,1-(2,6,6-TRIMETHYL-2-	245891.4

5	CYCLOHEXEN-1-YL) -	
	BUT-2-ENE-1-ONE, 1-(2,6,6-TRIMETHYL-3-CYCLOHEXEN-1-YL) -	243343.3
10	METHYL AMYL KETONE	239523.8
	CYCLOHEXENE, 1-METHYL-4-(1-METHYLETHYLIDENE)	234817.8
15	CITRODYLE	230257.8
	ALPHA ISO METHYLIONONE	218736.2
20	3,7 DIMETHYL-1,6-OCTADIEN-3-OL	218426.5
	METHYL OCTINE CARBONATE	216945.5
	ACETOPHENONE EXTRA	212908.8
25	4-PENTEN-2-OL, 3-METHYL-5-(2,2,3-TRIMETHYL-3-CYCLOPENTEN-1-YL)	212557.4
	OCTALACTONE, GAMMA	210058.7
30	INDOLE PURE	207042.8

[0049] The fragrances are present in an amount of 5 to 50%, preferably 10 to 30% with respect to the dry polymer, and 1.4 to 14%, preferably 3.5 to 10.5% with respect to a latex dispersion.

[0050] A typical nanoparticle of the invention comprises:

67% by weight of a copolymer phase consisting e.g. of 92% by weight of styrene, 2% by weight of divinyl benzene and 6% by weight of methacrylic acid,

ca. 30% by weight olfactive components,

0.6 to 3% by weight of the total composition of an emulsifier as stabiliser e.g. comprising SDS, Abex® 3594, Dowfax®, 2A1, Lexemul® AS and Mergital® LM4L.

[0051] The nanoparticles may be supplied or used suspended in water as latex or in solid dried form. The latex form does preferably contain not more than 70% by weight of the nanoparticles of the invention.

[0052] The effect of long lasting release of olfactive component is particularly pronounced when the latex is added to a conditioning composition. Significantly improved deposition, sustained release at room temperature and fast release at higher temperature are also found when the latex is added to rinsing water without softening components thus reducing the amount of small molecular cationic surfactants and the pollution of water.

[0053] The design of a latex for use in fabric care and hair conditioners constitutes a preferred embodiment of the present invention. Fabric care and hair conditioners comprise typically of cationic surfactants having a high affinity to fabrics and hair and providing desired softening action and esthetic shine.

[0054] The prior art teaches how to formulate perfumed particles in consumer products like liquid detergents and conditioners. A classical strategy involves the addition of particles bearing the same electrical charge as the principal surface active agent present in the product composition. Hence, particles disclosed so far for use in fabric care of hair conditioners are positively charged in order to increase their compatibility with the cationic actives present in the product.

Similarly, negatively charged particles are thought to be compatible with the anionic cleansing active present in liquid detergents. It has now been surprisingly found that the above restrictions do not apply to the nanoparticles of the present invention. By choosing suitable functional monomers and emulsifiers, nanoparticles bearing an opposite charge with respect to the principal surfactant of the end product, excellent controlled release properties can be gained, without affecting the stability of the end product. Hence, surprisingly, anionic hard, glassy, nanoparticles of the invention when added to a cationic liquid conditioner are specially suitable for imparting long lasting delivery of top notes on fabrics and hair over a long period of time and fast release of the olfactive component upon heating. Nanoparticles that are particularly suitable for use in fabric care and hair care conditioners are produced by polymerizing a mix containing 5 to 50%, preferably 10 to 30% of olfactive components, 40 to 95% styrene, 0 to 10% divinyl benzene, 0 to 10% (meth) acrylic acid and 0.5 to 3% anionic emulsifier by emulsion polymerization. The nanoparticles with sizes in the range of several hundred nanometers can be mixed directly with the fabric conditioner and deposited onto the fabrics. Soak test an olfactory evaluation confirms a sustained release of olfactive compounds during the drying stage and the storage.

[0055] Therefore, the strategy disclosed in the preferred embodiments of the present invention involves the use of negatively charged nanoparticles in consumer products containing cationic surfactants as principal surface active agents and positively charged nanoparticles in consumer products containing anionic surfactants as principal surface active agents. As shown in the examples below this strategy leads to products with excellent performance.

[0056] The surface potential of the nanoparticles is believed to control the stability of the nanoparticles in the end product. The surface potential of a colloid particle depends on a number of factors like (i) the amount of ionized chemical groups present on the surface, (ii) the nature of the emulsifier adsorbed on the particle and (iii) the amount of counterions present in the vicinity of the nanoparticle. If the partially ionizable groups consist of weak acids or weak bases, the surface potential will be also controlled by the pH of the dispersion medium. The surface potential of colloid particles is usually measured by measuring the so-called zeta-potential of the particles. A complete definition of zeta-potential can be found for example in (R.J. Hunter, "Zeta Potential in Colloid Science", Academic Press, London, 1981). The zeta-potential of particles in a diluted dispersion can be measured by electroosmosis and electrophoresis, whereas in concentrated dispersions, electrokinetic sonic amplitude measurements are preferred. It has been shown for optimal stability in end products containing electrically charged species, the zeta-potential should not exceed some critical values, which depend on the nature and composition of the end products. For typical fabric care and hair care conditioners containing up to 5% cationic surfactants and a pH of 2.5, the critical value of the zeta-potential is estimated to -35 mV. For typical liquid detergents or cleansing compositions containing up to 10% anionic surfactants and a pH of 8, this critical values will lie around +18 mV.

[0057] Significant enhancement of deposition means, a measurable increase of the fragrance concentration on the substrate measured by solvent or thermal extraction.

[0058] Significantly improved sustained release of the olfactive components means a perceivable fragrance concentration in the headspace surrounding the dry fabric after 5 days. The sustained release is checked after 24 hours, during 5 days by olfactive evaluation by at least 5 pannelists.

[0059] Significantly improved fast release at higher temperature of the olfactive components means an unambiguous increase of the concentration in the headspace surrounding the dry fabric during and after thermic treatment, as measured by thermic extraction and olfactive evaluation by at least 5 panelists.

[0060] The delivery system of the invention can be used in detergents or fabric conditioners, rinsing compositions for fabrics, body cleansing composition, hard surface cleaners, rinse off hair conditioners or cosmetic compositions. These compositions may be prepared by adding the delivery system to an unperfumed detergent or fabric conditioning composition, or to water as a rinsing composition for fabrics, or to rinse off hair conditioning or cosmetic composition. Further, the compositions of the invention may be perfume dispensers in form of a solid composition or applied to a solid carrier, or air fresheners.

[0061] The following examples show a preferred embodiment of the invention without limiting it.

Example 1

Preparation of a latex containing nanoparticles:

[0062] A pre-emulsion is prepared by mixing an aqueous phase, prepared by dispersing the surfactants Abex® 3594 (8g) and SDS (4g) in water (100g), with an organic phase containing styrene (276g), methacrylic acid (18g), divinylbenzene (6g) and the olfactive component (40g) (see Table 2, below). The aqueous and the organic phase are mixed, vortexed, homogenised (Ultraturrax® homogeniser) and flushed with nitrogen. The pre-emulsion has to be stable for more than 24 hours.

[0063] A 1ℓ reaction flask equipped with a stirrer, reflux condenser, thermometer and inlet tube for delivery from a peristaltic pump is placed in a water bath at 75 °C. During nitrogen rinsing, a first initiator I₁ (6 g Na₂S₂O₈ / 30 ml water)

EP 1 146 057 A1

is added dropwise into the reaction flask which contains 100 ml water, 0.3 g buffer (NaHCO_3), 0.5 g Abex® 3594 and a small amount of Iron(II) sulfate. After 30 minutes the pre-emulsion and a second initiator I_2 (3 g $\text{Na}_2\text{S}_2\text{O}_8$ / 60 ml water) are separately added dropwise into the reaction flask under stirring at 420 rpm, using peristaltic pumps over a period of about 120 minutes. After terminating the addition, the reaction mixture is stirred for further 30 minutes and the bath temperature is increased up to 88°C. Subsequently a third initiator I_3 (0.7 g $\text{Na}_2\text{S}_2\text{O}_8$ / 30 ml water) is added dropwise over a period of 30 minutes before the reaction mixture is cooled to room temperature. Finally the latex particles are filtered through a 150 micrometer sieve.

[0064] The olfactory performance of each sample is evaluated by a panel of at least 5 trained evaluators against that of a reference sample.

[0065] Example A: Variation of the methacrylic acid amount: 0 %, 2.6 %, 5.2 %, 7.8 % of the latex dispersion.

Table 3:

Olfactive performance for example A and general stability of the Polymerdispersion				
Methacrylic acid amount [%]	Olfactive Performance			Stability* of the Polymerdispersion
	24 h	48 h	> 72 h	
0	++	+	+	-
2.6	+++	++	+	+
5.2	+++	++	++	+
7.8	++	++	++	+

* Stability of the Polymerdispersion:
agglomeration after 1 week: -
no agglomeration after 1 week: +

[0066] Example B: Variation of the amount of perfume A and B: 5.9 %, 11.7 % of the latex dispersion.

Table 4:

Olfactive performance for example B		
perfume amount [%]	Olfactive perfume A	Performance perfume B
5.9	++	++
11.7	++	++

[0067] Example C: Variation of the amount of perfume C: 2.9 %, 5.9 %, 11.7 %, 17.7 % of the latex dispersion.

Table 5:

Olfactive performance for example C	
perfume amount [%]	Olfactive Performance
2.9	+
5.9	++
11.7	++
17.7	+

Example 2

Hair or fabric conditioner

[0068] 7 g of the conditioner base is mixed with various amounts of a latex comprising nanoparticles obtained according to Example 1. The amounts of polymer suspension are chosen in such a way that 1% with reference to the amount of conditioner of perfume composition is included. The mixture is intensively vortexed and homogenised by an Ultraturrax® homogeniser.

Example 3Assessment of storage stability

[0069] The stability of different mixtures of softener and latex dispersion are investigated over a period of one month. Depending on the composition of the latex dispersion, i.e. the amount of methacrylic acid, amount of fragrance, pH-value and temperature, some mixtures are becoming gel-like. The following table shows the stability for the latex of Example 1 in a dispersion of Example 2.

Table 6

Amount of softener	Amount of dispersion	1 day	4 days	10 days	1 month
7 g	1.200 g	Gel	Gel	Gel	Gel
7 g	1.008 g	Liquid	Gel	Gel	Gel
7 g	0.806 g	Liquid	Liquid	Liquid	Liquid
7 g	0.495 g	Liquid	Liquid	Liquid	Liquid
7 g	0.310 g	Liquid	Liquid	Liquid	Liquid

Example 4Rinse Test

[0070] 1.6 g of the fabric conditioner/latex mixture according to example 2 is diluted in 1 l of cold water. A dry cotton towel of 50 g is given into the mixture and washed at room temperature by rotating intensively for 10 minutes. Finally the towels are spin dried for 1 minutes and installed on a line dryer for 5 days. The olfactory performance of each sample is evaluated by a panel of at least 5 trained evaluators against that of a reference sample without nanoparticles as a function of time.

[0071] The result of the panel test release of perfume A from dry fabrics as a function of time is shown in figure 1.

Example 5Headspace analysis

[0072] The concentration of fragrances in the headspace surrounding towels handled, as in Example 4, with a fabric care conditioner with fragrance added (Reference) and with nanoparticles of the invention A added. 50 ml headspace were pumped through a Porapak® adsorber cartridge for one minute. The adsorbed fragrances are then extracted with 50 microliters hexane and the whole extract is injected into a large volume injection port of a gas chromatograph equipped with a capillary column. The area counts (arbitrary units) of each test substances are reported on Table 7.

Table 7

Perfume Compound	t _R	Area counts after 24 h	
		Reference	Sample
Amylacetat	8.82	0	730
Eucalyptol	9.60	27	6664
Dimethyloctenon	12.52	20	4086
Cyclal C	13.54	0	3266
Linalool	16.99	0	14
Aldehyde C12	19.33	67	984
Viridine	21.15	0	0
Terpineol	21.60	0	253
Benzylacetat	22.64	0	21
Alpha-ionone	26.52	0	238
Compound X	27.54	0	237

Table 7 (continued)

Perfume Compound	t_R	Area counts after 24 h	
		Reference	Sample
Phenyl ethyl alcohol	28.77	9	13
Diphenyloxide	31.66	30	747
Compound Y	32.28	11	130
Lilial	33.00	116	131

[0073] The results show clearly the remanence of the volatile top notes in the headspace of the towel after 24 hours drying.

Example 6

Determination of rest monomer

[0074] 3 g of polymerdispersion according to example 1 is weighted into a 10 ml head space vial and analyzed by a SPME (solid phase micro extraction) head space analysis at 35 °C, and compared to a sample doped with pure monomers.

[0075] According to example 1 a styrol content of 50 ppm is found.

Claims

1. Polymeric nanoparticles including olfactive molecules and having a glass temperature of $>50^{\circ}\text{C}$ obtainable by
 - a) continuously adding a liquid monomer component and an olfactive component to an aqueous solution of a first initiator comprising an emulsifier and distributing the added components in the aqueous solution to obtain a reaction mixture while starting polymerization of the monomer component in the reaction mixture at a first temperature and, while continuing the addition of the liquid monomer component and the olfactive component, adding a second initiator dropwise to the reaction mixture while maintaining the first temperature
 - b) after terminating the addition of the liquid monomer component and the olfactive component increasing the temperature of the reaction mixture to a second temperature and dropwise adding a third initiator.
2. Nanoparticles according to claim 1 having a particle size of $<3 \cdot 10^{-6}$ m preferably between 250 and $500 \cdot 10^{-9}$ m.
3. Nanoparticles according to one of the preceding claims comprising 5 to 50%, preferably up to 30%, by weight of olfactive components.
4. Nanoparticles according to one of the preceding claims comprising a body part and an outer layer or coating part whereby the two parts are formed by different polymers.
5. Nanoparticles according to one of the preceding claims with characteristics changing from centre to the surface of the particles.
6. Nanoparticles according to one of the preceding claims selected from polystyrene and acrylic polymers.
7. Nanoparticles according to one of the preceding claims comprising cross linked polymer.
8. Nanoparticles according to one of the preceding claims comprising a template polymer including the olfactive components.
9. Nanoparticles according to one of the preceding claims whereby up to 70% by weight of nanoparticles are suspended in a water phase as latex.

10. Nanoparticles according to any of the proceeding claims comprising residual monomer of 100 ppm or less of the principal monomer compound.

11. Nanoparticles according to any of the preceeding claims comprising 50 ppm or less of styrene of a styrene/acrylic acid monomer whereby styrene is the principal monomer component.

12. Method of preparing nanoparticles including olfactive molecules and having a glass temperature of $>50^{\circ}\text{C}$ by

a) continuously adding a liquid monomer component and an olfactive component to an aqueous solution of a first initiator comprising an emulsifier and distributing the added components in the aqueous solution to obtain a reaction mixture while starting polymerization of the monomer component in the reaction mixture in the presence at a first temperature and, while continuing the addition of the liquid monomer component and the olfactive component, adding a second initiator dropwise to the reaction mixture while maintaining the first temperature

b) after terminating the addition of the liquid monomer component and the olfactive component increasing the temperature of the reaction mixture to a second temperature and dropwise adding a third initiator.

13. Method according to claim 12 whereby the olfactive component is a mixture of olfactive components and the composition of the mixture is varied during the addition thereof.

14. Method according to claim 12 or 13, whereby the temperature is varied during addition of the components.

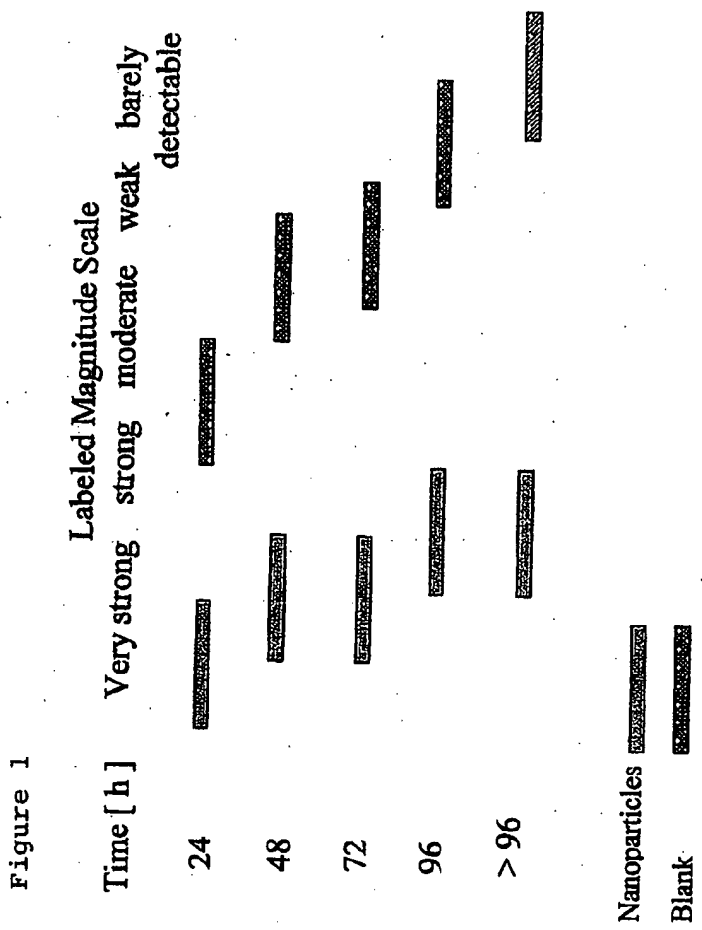
15. Delivery system for olfactive components comprising nanoparticles according to one of the claims 1 to 12.

16. Delivery system according to claim 15 comprising an anionic or cationic surface active agent and a nanoparticle having an electrical charge opposite to the electrical charge of the surface active agent.

17. Delivery system according to claim 16 comprising a cationic surface active agent and anionic nanoparticles.

18. Delivery system according to one of the claims 15 to 17 being a hair or fabric conditioner.

19. Delivery system according to one of the claims 15 to 17 being liquid detergents for fabrics, hard surface cleaners, skin care products, cosmetics.





European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 00 10 8313

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
A	EP 0 839 902 A (BASF) 6 May 1998 (1998-05-06)		C08F2/44 C11D17/00 A61K7/46
D,A	EP 0 617 051 A (ALLIED COLLOIDS LTD.) 28 September 1994 (1994-09-28)		A61K7/48 A61K7/06 A61K7/00
A	DE 967 860 C (C.H. BOEHRINGER SOHN)		C11D3/50
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
			C08F C11D
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 7 November 2000	Examiner Cauwenberg, C
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons a : member of the same patent family, corresponding document	
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document			

EPO FORM 1503 (02.02.92) (Pct/Cv)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 00 10 8313

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on. The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

07-11-2000

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
EP 839902	A	06-05-1998	DE	19645024 A	07-05-1998
			US	5972508 A	26-10-1999
EP 617051	A	28-09-1994	AT	121427 T	15-05-1995
			AU	2269488 A	31-03-1989
			AU	618553 B	02-01-1992
			DE	3853604 D	24-05-1995
			DE	3853604 T	31-08-1995
			DK	188589 A	20-06-1989
			EP	0305139 A	01-03-1989
			EP	0330692 A	06-09-1989
			ES	2070854 T	16-06-1995
			FI	891918 A	21-04-1989
			WO	8901948 A	09-03-1989
			JP	2500598 T	01-03-1990
			NO	891651 A	20-06-1989
DE 967860	C		NONE		

EPO FORM PUS8

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82